Micromechanical Assessment of Al/Y-substituted NASICON Solid Electrolytes

Juliane Franciele Nonemacher¹, Sahir Naqash^{1,2}, Frank Tietz^{1,2}, Jürgen Malzbender^{1*}

¹ Forschungszentrum Jülich GmbH, Institute of Energy and Climate Research, 52425 Jülich,

Germany

² Forschungszentrum Jülich GmbH, Helmholtz-Institute Münster, 52425 Jülich, Germany

ABSTRACT

Al/Y-substituted NASICON solid electrolytes are promising for novel solid state batteries. The

solid solutions $Na_{1+2x}Al_xY_xZr_{2-2x}(PO_4)_3$ (NAYZPx) and $Na_{3+2x}Al_xY_xZr_{2-2x}(SiO_4)_2(PO_4)$

(NAYZSiPx) that crystallize in this structure are characterized by depth-sensitive indentation

technique to determine their micro-mechanical properties. NAYZPx is rhombohedral, whereas

NAYZSiPx changes from monoclinic to rhombohedral phase depending on substitutional level

and temperature. For rhombohedral NAYZPx elastic moduli are in the range ~ 72 to 82 GPa and

hardness values are in the range ~ 4.8 to 5.8 GPa, whereas for NAYZSiPx the elastic moduli are

with the range ~ 72 to 88 GPa and hardness ~ 5.6 to 7.6 GPa, respectively. The fracture

toughness values for both systems, as determined by Vickers indentation, are rather independent

of the applied load being with the range 1.30 to 1.58 MPa·m^{0.5}. Cracks typically show a mixed

intergranular/transgranular crack mode.

Keywords: Solid electrolyte; NASICON; Mechanical properties; Elastic modulus; Hardness;

Fracture Toughness;

* Corresponding author: j.malzbender@fz-juelich.de, tel.: ++ 49 2461 6964, fax: ++ 49 2461 613699

1

1. Introduction

Structural stability and ionic conductivity makes NASICON (Na⁺ Super Ionic Conductor) materials promising for the application as solid electrolytes for solid-state sodium batteries. Originally the system Na_{1+x}Zr₂P_{3-x}Si_xO₁₂ ($0 \le x \le 3$) has a minimum Na⁺-ion resistivity of ≤ 5 Ω -cm at 300 °C, which is similar as that of the conventional sodium-ion conductor β'' -alumina [1,2]. The crystal structure is rhombohedral ($R\overline{3}c$) and tends, depending on composition, towards a monoclinic distortion. There are several studies on synthesis, substitution and conductivity in literature (e.g. [3-6]); however, very few reports deal with the mechanical properties [7,8].

In addition to high ionic conductivity, a solid electrolyte should exhibit necessary mechanical reliability. Therefore, it is necessary to gain more knowledge on mechanical performance to warrant safety and reliability. It has been reported that Na-S batteries with β/β'' -alumina revealed failure due to Na filaments that have propagated through the solid electrolyte [9-11]. In fact, the fracture toughness is a crucial parameter for crack growth and hence potentially for short-circuiting in solid-state batteries [12,13].

The indentation is a frequently used method to derive fracture toughness values, which requires only a small materials' volume and simple specimens' preparation. Cracks are induced by a sharp stylus, mostly Vickers or Berkovich type, and the length of the resultant cracks that are induced above a critical load is measured. For the fracture toughness calculation, due to the relationship to elastic and plastic volume, knowledge of elastic modulus and hardness is a prerequisite [14-15].

Furthermore, the elastic modulus is also a very important parameter when designing a solid-state battery [16-22]. The solid electrolytes should deform elastically and accommodate internal stresses and should be able to assure a good electrode/electrolyte interface [23].

Thus, the purpose of this study is the determination of the mechanical properties of NASICON solid electrolyte with double substitution of Zr^{4+} with Al^{3+} and Y^{3+} [24]. Elastic modulus, hardness and fracture toughness are determined via indentation testing as function of the applied load, and potential correlations with substitution level and microstructural features are discussed.

2. Experimental

The NASICON solid solutions $Na_{1+2x}Al_xY_xZr_{2-2x}(PO_4)_3$ (NAYZPx, $0 \le x \le 0.25$) and $Na_{3+2x}Al_xY_xZr_{2-2x}(SiO_4)_2(PO_4)$ (NAYZSiPx, $0 \le x \le 0.3$) were prepared using the Pechini method [5]. The obtained gels were fired at 600 °C for 3 h and amorphous precursors were formed and calcined at 800 °C for 12 h. The calcined powders were milled and pressed into pellets. The NAYZPx pellets were sintered at 1200 °C for 10 h in air, whereas the NAZYSiPx samples were exposed to a two-step sintering process, first at 1200 °C for 0.5 h and then at 1150 °C for 5 h in air. More details of synthesis can be found in [24]. The relative density of the specimens was within the range 85 to 93%.

Phase purity of the sintered specimens was controlled by X-ray diffraction. Rietveld refinement was performed to calculate the lattice parameters [24]. The microstructure was characterized using cross-sections by field emission scanning electron microscopy (FE-SEM; Zeiss Merlin). The grain size distribution was determined using the software AnalySIS Pro based on the circular intercept segment mode. Electrical conductivity was measured using an impedance spectroscopy

system (Biologic, VMP300) from -30 °C to 100 °C in the frequency range 4 MHz to 300 Hz using sputtered gold as blocking electrodes.

The indentation method was used to evaluate hardness (H), elastic modulus (E) and fracture toughness (K_{IC}). Cross-sections of pellets were embedded in epoxy resin and finely polished via alumina suspension. A matrix of 5 × 5 indentation imprints was performed for each load. H and E values based on Oliver-Pharr procedure [25] were obtained as function of the applied loads, i.e. at 50 mN, 100 mN and 1000 mN, using a Fischer HC100 machine with a Vickers diamond tip, see also [26,27]. Poisson's ratio for all compositions was assumed to be 0.27 [7].

The fracture toughness, K_{IC} , was derived from the cracks induced from the corners of indentations after loading to 1 N, 2.94 N, 4.9 N and 9.81 N. The dependence of cracks lengths and hence fracture toughness on the load was checked. Due to the load limitations, the Fischer HC100 machine was used only for 1 N, while for higher loads a Micromet, Buehler Ltd., was employed. K_{IC} was derived on the basis of a Palmqvist crack system, since it was observed that crack lengths (l) and half-diagonals (a) were in the range $0.25 \le l/a \le 2.5$. The used Niihara equation reads [14]:

$$K_{IC} = 0.035 \left(\frac{H.a^{0.5}}{\phi}\right) \left(\frac{E \phi}{H}\right)^{0.4} \left(\frac{a}{l}\right)^{0.5} \tag{1}$$

where E is the elastic modulus, H the hardness and ϕ a constraint factor (typically 3 is used). Elastic modulus and hardness obtained at 50 mN were used to calculate K_{IC} , since these load load values are assumed to be representative for the intrinsic material's properties. Microstructure and

indentation imprints were visualized by SEM, permitting also measurement of the crack lengths, which was carried out immediately after the impression.

3. Results and discussion

From the X-ray diffraction patterns it was identified that the NAYZPx are in rhombohedral phase with space group $R\overline{3}c$, and some minor secondary phase content (YPO₄, AlPO₄, ZrO₂) for x being in the range of 0 to 0.15 and x equal to 0.25. In case of the sample with x = 0.20 the amount of YPO₄ is unusual high, about 10 wt.%, and in case of x > 0.25 a continuous increase in YPO₄ and Na₃Y₂(PO₄)₃ with increasing x indicates that the solid solution is at its limit [24]. For NAYZSiPx a monoclinic structure was found in the range $0 \le x \le 0.1$ and a transition to a rhombohedral phase for 0.2 < x < 0.3, with a small amount of monoclinic ZrO₂ detected in all compositions. Table 1 summarizes space group, lattice parameters and ionic conductivity (σ) at room temperature for all samples, where a_{hex} and c_{hex} are the hexagonal lattice constants and a_{r} is the rhombohedral lattice constant.

Table 1- Composition, space group, lattice parameters and room temperature ionic conductivity for the $Na_{1+2x}Al_xY_xZr_{2-2x}(SiO_4)_y(PO_4)_3$ system.

Composition	Space group	a _{hex} (Å)	Chex (Å)	a _r (Å)	σ (S cm ⁻¹)
NaZr ₂ (PO ₄) ₃	R3c	8.805	22.776	9.137	3.1 · 10-7
$Na_{1.1}Al_{0.05}Y_{0.05}Zr_{1.9}(PO_4)_3$	R3c	8.812	22.779	9.140	$1.1 \cdot 10^{-7}$
$Na_{1.2}Al_{0.1}Y_{0.1}Zr_{1.8}(PO_4)_3$	R3c	8.816	22.789	9.144	$5.0 \cdot 10^{-6}$
$Na_{1.3}Al_{0.15}Y_{0.15}Zr_{1.7}(PO_4)_3$	R3c	8.824	22.805	9.151	$6.0 \cdot 10^{-6}$
$Na_{1.4}Al_{0.2}Y_{0.2}Zr_{1.6}(PO_4)_3$	R3c	8.830	22.820	9.157	$6.2 \cdot 10^{-6}$
$Na_{1.5}Al_{0.25}Y_{0.25}Zr_{1.5}(PO_4)_3$	R3c	8.849	22.825	9.165	$1.2 \cdot 10^{-6}$
$Na_3Zr_2(SiO_4)_2(PO_4)_3$	C2/c	9.032*	23.042*	9.284	$1.7 \cdot 10^{-3}$
$Na_{3.1}Al_{0.05}Y_{0.05}Zr_{1.9}(SiO_4)_2(PO_4) \\$	C2/c	9.047*	23.021*	9.283	$1.6 \cdot 10^{-3}$
$Na_{3.2}Al_{0.1}Y_{0.1}Zr_{1.8}(SiO_4)_2(PO_4)$	C2/c	9.052*	22.948*	9.264	$1.3 \cdot 10^{-3}$
$Na_{3.4}Al_{0.2}Y_{0.2}Zr_{1.6}(SiO_4)_2(PO_4)$	R3c	9.096	22.741	9.222	$1.4 \cdot 10^{-3}$
$Na_{3.6}Al_{0.3}Y_{0.3}Zr_{1.4}(SiO_4)_2(PO_4)$	R3c	9.110	22.671	9.207	$4.6 \cdot 10^{-4}$

^{*:} transformed from the monoclinic system, values are only given for the sake of comparison.

Figure 1 presents as examples the microstructures of NAYZPx with x = 0.10, 0.20 and 0.25 and NAYSiZPx with x = 0, 0.05 and 0.1. The microstructures are rather homogeneous with small grain sizes, consistent and as expected for such powders obtained via the Pechini method.

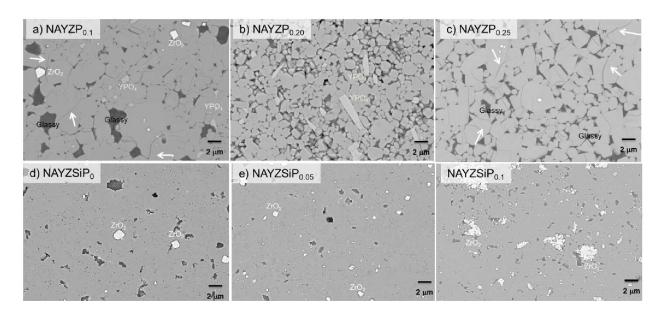


Figure 1- SEM images of polished cross-section of different NAYZPx and NAYZSiPx. Granular micro-cracks appear for non Si-substituted compositions NAYZPx as indicated by arrows. White ZrO₂ and grey YPO₄ secondary phases are also visible. The dark region between the grains is glassy phase [24,28,29].

For the NAYZPx system, compositions with x=0 and x=0.05 did not fully densify at the sintering temperature and relative densities of only 52% and 75% were achieved. For $0.1 \ge x \ge 0.25$, the relative density of the samples varied between 87% and 91%. However, overall microstructures were very similar. For NAYZPx with x=0.1 and 0.25, micro-cracks were observed and YPO $_4$ and ZrO $_2$ as secondary phase, highlighted in Figure 1.

Comparing the NAYZSiPx specimens, relative densities varied from 85-93% with rather similar microstructures for all compositions. In addition, the amount of ZrO_2 secondary phase increased when x was increased, as can be seen in Figure 1.

Elastic modulus and hardness - NAYZPx

Figure 2 presents elastic modulus and hardness as a function of the applied load for the NAYZPx system. Overall, E and H decrease with increasing the load. The decrease with load (depth) can be associated with a larger contribution of grain boundaries and porosity. At a low load of 50 mN the penetration depths are around 600 nm, so the properties for low loads are assumed to be representative of the materials intrinsic properties.

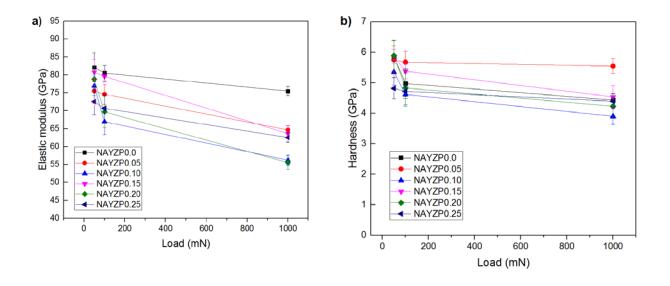


Figure 2- Elastic modulus and hardness as function of load for NAYZPx in (a) and (b).

Considering low load values, for the NAYZPx series, the unsubstituted material (x = 0) possesses the highest elastic modulus of 82 ± 4 GPa and with increasing substitutional level values decrease

to 73 \pm 4 GPa. Hardness values range from 5.8 \pm 0.5 GPa for the unsubstituted to 4.8 \pm 0.3 GPa for the material with maximum amount of substitution.

The values at low load are used to assess the compositional impact, see Figure 3. The unsubstituted material shows the highest values of mechanical parameters, followed by a reduction with increasing the Al/Y content to 0.05 and 0.1, and then a rise of the mechanical properties at 0.2 with again a diminution of E and H with increasing the substitution level to the maximum of 0.25. In case of x > 0.25 the continuous increase in YPO₄ and Na₃Y₂(PO₄)₃ with increasing x that indicated that the solid solution was at its limit [24], might have led to reduced values.

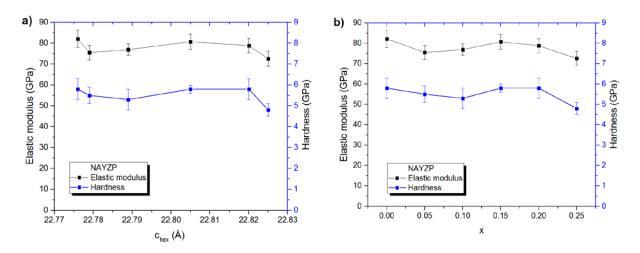


Figure 3- Elastic modulus and hardness of NAYZPx as function of a) hexagonal lattice parameter and b) composition *x*.

The elastic modulus is related to lattice parameters and atomic bonding. It can be expected that the larger the lattice parameter for the same crystal structure, the weaker the atomic bonding and therefore the lower the elastic modulus. In order to verify this relationship, E and H were plotted

as function of hexagonal lattice parameter c (data Table 1). In fact, plotting the data as a function of the rhombohedral lattice constant, in order to consider also the hexagonal a-axis, would lead to an equivalent representation since hexagonal and rhombohedral lattice parameter are almost linear proportional, see Table 1.

The calculated elastic modulus of NaZr₂(PO₄)₃ with space group $R\bar{3}c$ is 120.9 GPa [30], which is much higher than the value obtained experimentally for the NAYZPx in this work. This difference might be attributed to fact that the measured values are for polycrystalline samples and depend on various additional factors such as local porosity and grain boundaries, in addition the theoretical value is normally calculated at 0 K and a decrease with increasing temperature can be expected.

Elastic modulus and hardness - NAYZSiPx

Figure 4 presents elastic modulus and hardness as a function of the applied load for the NAYZSiPx system. Again, E and H decrease with increasing the load. The mechanical properties reveal a stronger dependency on the load compared to the NAYZPx series, see Figure 2. For reasons leading to a decrease with increasing load see previous section, the properties for low loads are assumed to be representative of the materials intrinsic properties.

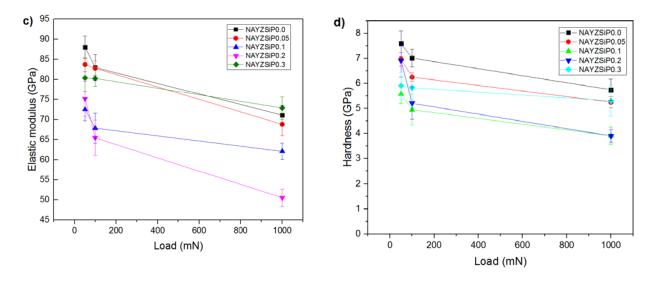


Figure 4- Elastic modulus and hardness as function of load for NAYZSiPx in (a) and (b).

For the NAYZSiPx, again, the unsubstituted material has the highest elastic modulus of 88 ± 3 GPa and the lowest is found for NAYZSiP0.1 with 73 ± 3 GPa. Elastic modulus and hardness of NAYSiPx are shown in Figure 5 as function of hexagonal lattice parameter and composition x. Overall, behavior is rather complex with maximum values for low x (high c_{hex}). Note, for NAYZSiPx a monoclinic structure was found in the range $0 \le x \le 0.1$ and a transition to a rhombohedral phase for 0.2 < x < 0.3, with a small amount of monoclinic ZrO $_2$ was detected in all compositions, where especially the monoclinic structure might be responsible for the higher data at low x.

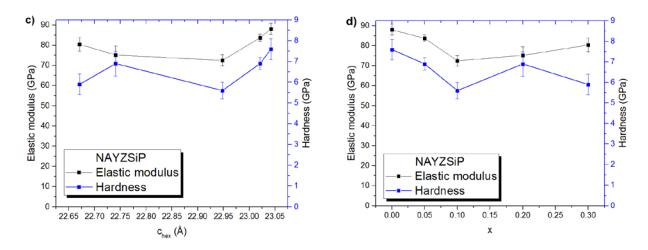


Figure 5- Elastic modulus and hardness of NAYSiPx as function of a) hexagonal lattice parameter and b) composition *x*.

Fracture toughness - NAYZPx

For NAYZPx the ratio l/a is in the range of 0.7 to 1.3, fulfilling the requirement of $0.25 \le l/a \le 2.5$ for Palmqvist crack system. Figure 6 presents the indentation marks on the surface of NAYZP0.05 and NAYZP0.10 for different loads. It can be clearly seen that the cracks do not propagate along a straight line, but have a tortuous path, showing typically a combined intergranular / transgranular fracture mode. This can be related to the week bonding between grains. Furthermore, due this complex non-linear crack path, it is very complicated to determine precisely the crack length. For the applied load of 1 N, it was not possible to determine the crack length due the complex interaction of crack shape with microstructure; for the other loads the crack length was measured using the software AnalySIS Pro.

Crack lengths and fracture toughness values are compiled in Figure 7. The slight increase of K_{IC} for some compositions at higher loads might be a result of start of fracture path changes from Palmqvist to median / radial.

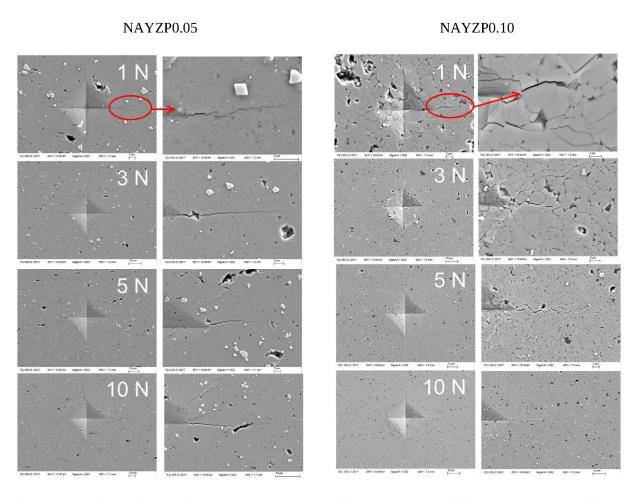


Figure 6- Indentation marks on NAYZP0.05 and NAYZP0.10 with 1 N, 3 N, 5 N and 10 N; fracture path is highlighted in red.

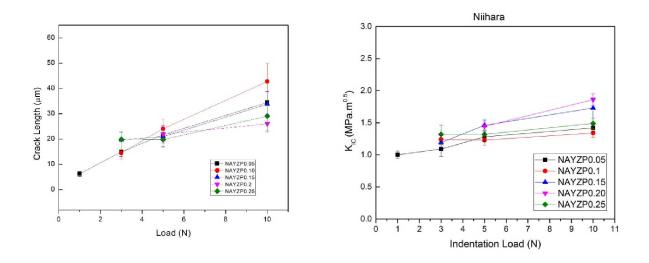


Figure 7- Crack length (a) and fracture toughness (b) of NAYZPx as function of load.

For NAYZP0.0 no measurable cracks could be observed for any load due to the complex crack shape, the same happened for other compositions at loads of 1 and 3 N. Therefore fracture toughness was not calculated in these cases. At 5 N, values could be obtained and corresponding values are shown in Figure 8.

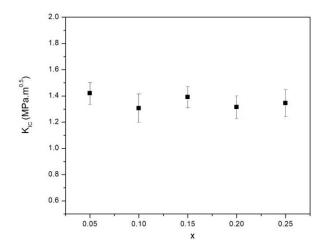


Figure 8- Fracture toughness of NAYZPx for 0.05 < x < 0.3 (load 5 N).

The values showed no significant variation with composition, being 1.3 ± 0.1 MPa·m^{0.5} for NAYZP0.1 and 1.4 ± 0.1 MPa·m^{0.5} for NAYZP0.05, overall with a similar dependence as for the elastic modulus. Hence, the composition has small influence on the fracture toughness.

Fracture toughness - NAYZSiPx

A similar investigation was performed on the series NAYZSiPx. Figure 9 shows typical indentation marks for NAYZSiP0.2 and NAYZP0.3. Values of $K_{\rm IC}$ as function of load are shown in Figure 9. For the applied load of 1 N, it was not possible to measure crack lengths, therefore

images are not shown here. A slight increase of fracture toughness above 10 N might have been a result of initial changes from Palmqvist to median / radial crack type.

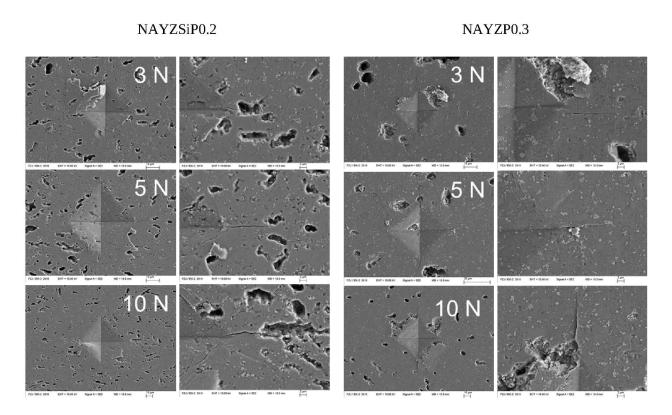


Figure 9- Indentation marks for NAYZSiP0.2 and NAYZP0.3 with 3 N, 5 N and 10 N.

The impressions at 5 N yielded well developed cracks, therefore the fracture toughness at 5 N was used to assess compositional effects, see Figure 10. The $K_{\rm IC}$ values slightly increase up to x = 0.1 and then decrease towards x = 0.3, varying from 1.85 \pm 0.15 MPa·m^{0.5} for NAYZSiP0.1 to 1.49 \pm 0.04 MPa·m^{0.5} for NAYZSiP0.3.

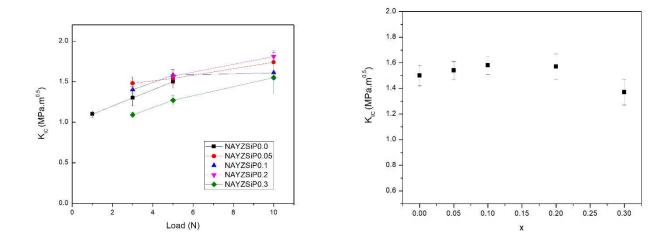


Figure 10- Fracture toughness of NAYZSiPx with 0.0 < x < 0.3 obtained at 1, 3, 5, and 10 N in (a) and at 5 N in (b) as a function of substitutional level.

4. Conclusions

Mechanical properties of NASICON-type materials with double substitution of Zr with Al and Y were assessed via depth-sensitive indentation. The dependency of elastic modulus, hardness and fracture toughness on applied load and composition was investigated.

Elastic modulus of NAYZPx showed a dependency on applied load, whereas hardness remained constant. Elastic moduli derived from low load data decreased from 82 ± 4 GPa for NaZr(PO₄)₃ (NAYZPx) to 73 ± 4 GPa for Na_{3.5}Al_{0.25}Y_{0.25}Zr_{1.5}(PO₄)₃ and hardness from 5.8 ± 0.5 GPa and 4.8 ± 0.3 GPa. Fracture toughness evaluated at 5 N ranged from 1.3 ± 0.1 MPa·m^{0.5} for NAYZP0.1 to 1.4 ± 0.1 MPa·m^{0.5} for NAYZP0.05.

For NAYZSiPx, Na₃Zr₂(SiO₄)₂(PO₄) possessed the highest elastic modulus (88 \pm 3 GPa) and the lowest was obtained for Na_{3.2}Al_{0.1}Y_{0.1}Zr_{2.8}(SiO₄)₂(PO₄) (73 \pm 3 GPa), both crystallizing in monoclinic structure. Rhombohedral NAYZSiP0.2 and NAYZSiP0.3 have rather similar values

within the same range of uncertainty of 75 ± 4 GPa and 80 ± 3 GPa, respectively. Hardness varies from 7.6 ± 0.5 GPa for Na₃Zr₂(SiO₄)₂(PO₄) to 5.6 ± 0.4 GPa for Na_{3.2}Al_{0.1}Y_{0.1}Zr_{2.8}(SiO₄)₂(PO₄). Focusing on the influence of composition on fracture toughness, the values at 5 N tended to remain rather constant up to NAYZSiP0.2 and slightly decreased to 1.4 ± 0.1 MPa·m^{0.5} for NAYZSiP0.3.

Acknowledgements

This work was supported by the National Council for Scientific and Technological Development under grant number: 232838/2014-3 (CNPq)-Brazil. The authors would also like to especially thank to Dr. E. Wessel, M. Ziegner for their cooperation with structural characterization, as well as to Prof. Dr. M. Krüger, Prof. Dr. L. Singheiser and Prof. Dr. O. Guillon for support.

References

- 1. J.B. Goodenough, H.Y.P. Hong, J.A. Kafalas, Fast Na⁺-ion transport in skeleton structures, Mater. Res. Bull. 11 (1976) 203-220.
- 2. H.Y.P. Hong, Crystal structures and crystal chemistry in the system Na_{1+x}Zr₂Si_xP_{3-x}O₁₂, Mater. Res. Bull. 11 (1976) 173-182.
- 3. M. Guin, F. Tietz, O. Guillon, New promising NASICON material as solid electrolyte for sodium-ion batteries: Correlation between composition, crystal structure and ionic conductivity of Na_{3+x}Sc₂Si_xP_{3-x}O₁₂, Solid State Ionics 293 (2016) 18-26.
- 4. O. Bohnke, S. Ronchetti, D. Mazza, Conductivity measurements on NASICON and NASICON-modified materials, Solid State Ionics 122 (1999) 127-136.
- 5. S. Naqash, Q. Ma, F. Tietz, O. Guillon, Na₃Zr₂(SiO₄)₂(PO₄) prepared by a solution-assisted solid state reaction, Solid State Ionics 302 (2017) 83-91.

- 6. T. Takahashi, K. Kuwabara, M. Shibata, Solid-state ionics conductivities of Na⁺ ion conductors based on NASICON. Solid State Ionics 1 (1980) 163-175.
- 7. Z. Deng, Z. Wang, I.-H. Chu, J. Luo, S.P. Ong, Elastic properties of alkali superionic conductor electrolytes from first principles calculations, J. Electrochem. Soc. 163 (2016) A67-A74.
- 8. A.V. Virkar, G.J. Tennenhouse, R.S. Gordon, Hot-pressing of Li₂O-stabilized β"-Alumina. J. Am. Ceram. Soc. 57 (1974) 508-508.
- 9. X. Lu, G. Xia, J.P. Lemmon, Z. Yang, Advanced materials for sodium-beta alumina batteries: Status, challenges and perspectives, J. Power Sources 195 (2010) 2431-2442.
- 10. M. Ohshima, A. Kobayashi, A. Yoshida, H. Tsuji, Verification of the fracture mechanism of β″-aluminas caused by large current density, Electrical Engineering in Japan 111 (1991) 11-23.
- L. Viswanathan, A.V. Virkar, Crack-propagation rates in β"-alumina, J. Am. Ceram. Soc.
 66 (1983) 159-162.
- 12. R. Raj, J. Wolfenstine, Current limit diagrams for dendrite formation in solid-state electrolytes for Li-ion batteries, J. Power Sources 343 (2017) 119-126.
- 13. E. J. Cheng, A. Sharafi, J. Sakamoto, Intergranular Li metal propagation through polycrystalline Li_{6.25}Al_{0.25}La ₃Zr₂O₁₂ ceramic electrolyte, Electrochim. Acta 223 (2017) 85-91.
- 14. K. Niihara, A fracture mechanics analysis of indentation-induced Palmqvist crack in ceramics, J. Mater. Sci. Lett. 2 (1983) 221-223.
- 15. G.R. Anstis, P. Chantikul, B.R. Lawn, D.B. Marshall, A critical evaluation of indentation techniques for measuring fracture toughness: I, direct crack measurements, J. Am. Ceram. Soc. 64 (1981) 533-538.
- 16. A.G. Evans, E.A. Charles, Fracture toughness determinations by indentation, J. Am. Ceram. Soc. 59 (1976) 371-372.

- 17. X. Hu, G. Yan, X. Cheng, J. Malzbender, W. Qiang, B. Huang, Electrochemical and mechanical stability of Li_xLa_{0.557}TiO_{3-δ} perovskite electrolyte at various voltages, J. Am. Ceram. Soc. 102 (2019) 1953-1960.
- 18. X. Hu, X. Cheng, S. Qin, G. Yan, J. Malzbender, W. Qiang, B. Huang, Mechanical and electrochemical properties of cubic and tetragonal Li_xLa_{0.557}TiO₃ perovskite oxide electrolytes, Ceram. Int. 44 (2018) 1902-1908.
- 19. G. Yan, J.F. Nonemacher, H. Zheng, M. Finsterbusch, J. Malzbender, M. Krüger, An investigation on strength distribution, subcritical crack growth and lifetime of the lithium-ion conductor Li₇La₃Zr₂O₁₂, J. Mater. Sci. 54 (2019) 5671-5681.
- 22. A.-N. Wang, J.F. Nonemacher, G. Yan, M. Finsterbusch, J. Malzbender, M. Krüger, Mechanical properties of the solid electrolyte Al-substituted Li₇La₃Zr₂O₁₂ (LLZO) by utilizing micro-pillar indentation splitting test, J. Euro. Ceram. Soc. 38 (2018) 3201-3209.
- 21. J.F. Nonemacher, C. Hüter, H. Zheng, J. Malzbender, M. Krüger, R. Spatschek, M. Finsterbusch, Microstructure and properties investigation of garnet structured Li₇La₃Zr₂O₁₂ as electrolyte for all-solid-state batteries, Solid State Ionics 321 (2018) 126-134.
- 22. G. Yan, S. Yu, J.F. Nonemacher, H. Tempel, H. Kungl, J. Malzbender, R.-A. Eichel, M. Krüger, Influence of sintering temperature on conductivity and mechanical behavior of the solid electrolyte LATP, Ceram. Int. 45 (2019) 14697-14703.
- 23. C. Monroe, J. Newman, The impact of elastic deformation on deposition kinetics at lithium/polymer interfaces, J. Electrochem. Soc. 152 (2005) A396-A404.
- 24. S. Naqash, F. Tietz, O. Guillon, Synthesis and characterization of equimolar al/y-substituted NASICON solid solution $Na_{1+2x+y}Al_xY_xZr_{2-2x}Si_yP_{3-y}O_{12}$, Solid State Ionics 319 (2018) 13-21.

- 25. W.C. Oliver, G.M. Pharr, An improved technique for determining hardness and elastic modulus using load and displacement sensing indentation experiments, J. Mater. Res. 7 (2011) 1564-1583.
- 26. J. Malzbender, Comment on hardness definitions, J. Euro. Ceram. Soc. 23 (2003) 1355-1359.
- 27. T. Wakui, J. Malzbender, R.W. Steinbrech, Surf. Coat. Technol. 200 (2006) 5419-5426.
- 28. S. Naqash, D. Sebold, F. Tietz, O. Guillon, Microstructure–conductivity relationship of Na₃Zr₂(SiO₄)₂(PO₄) ceramics. J. Am. Ceram. Soc. 102 (2019) 1057-1070.
- 29. S. Naqash, F. Tietz, E. Yazhenskikh, M. Müller, O. Guillon, Impact of sodium excess on electrical conductivity of Na₃Zr₂Si₂PO₁₂ + x Na₂O ceramics, Solid State Ionics, 336 (2019) 57-66.
- 30. Z. Deng, Z. Wang, I.-H. Chu, J. Luo, S. Ong, Elastic properties of alkali superionic conductor electrolytes from first principles calculations, J. Electrochem. Soc. 163 (2016) A67-A74.